

1 **Supplementary materials**

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3 Table 1: Conversion rates of aromatics in single, binary and ternary mixtures under different  
4 cultivation conditions. The corresponding generation times are given in the last column.

5 B: benzene T: toluene FB: fluorobenzene CB: chlorobenzene

Cultivation substrate / test substrate(s)	Substrate specific transformation rate [mg C/(L·h·OD)]				Transformation rate total [mg C/L·h·OD]	Generation time g [h]
	B	T	FB	CB		
B / B	121.4	-	-	-	121.4	6.8
B / T	-	178.2	-	-	178.2	5.0
B / FB	-	-	81.2	-	81.2	14.2
B / B + T	56.2	63.2	-	-	119.4	6.6
B / B + FB	24.5	-	51.0	-	75.4	12.1
B / B + FB + T	24.9	41.8	20.2	-	86.9	8.7
T / B	53.4	-	-	-	53.4	8.3
T / T	-	89.6	-	-	89.6	4.8
T / FB	-	-	40.7	-	40.7	16.4
T / CB	-	-	-	51.4	51.4	6.4
T / B + FB	38.8	-	26.7	-	65.5	5.5
T / B + T	29.4	40.9	-	-	70.3	5.6
T / FB + T	-	41.1	32.7	-	73.8	8.5
T / T + CB	-	25.0	-	20.2	45.1	12.6
T / B + FB + T	25.0	41.5	21.1	-	87.5	8.9
T / B + T + CB	61.8	77.4	-	50.1	189.4	13.5
FB / FB + T	-	50.0	31.7	-	81.7	6.2
FB / B + FB + T	25.7	31.9	21.5	-	79.1	7.4

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1 Table 2: Specific oxygen uptake activities for the initial enzyme(s) of cells of FLU100 pre-grown on  
 2 fluorobenzene, toluene or benzene. The value in parenthesis represents the number of  
 3 independent batch cultures tested twice. The first value describes the average value of  
 4 oxygen uptake while the second one describes the variation. The absolute activity for  
 5 fluorobenzene grown cells was  $465 \pm 77.2$  units,  $623.8 \pm 243.7$  units for toluene and  $629.4$   
 6  $\pm 190.1$  units for benzene. One unit of oxygenase activity was defined as the conversion  
 7 of  $1 \mu\text{g O}_2 \cdot \text{L}^{-1} \cdot \text{min}^{-1} \cdot \text{OD}^{-1}$ .

	Specific activity fluorobenzene (7) [%]	Specific activity toluene (11) [%]	Specific activity benzene (7) [%]
<b>Mono-halogen aromatics</b>			
fluorobenzene	$62.5 \pm 11.2$	$24.3 \pm 5.8$	$28.3 \pm 10.6$
chlorobenzene	$89.9 \pm 10.4$	$63.8 \pm 12.2$	$125.4 \pm 27.6$
bromobenzene	$92.7 \pm 12.6$	$60.6 \pm 6.1$	$119.0 \pm 18.0$
2-chlorotoluene	$-1.5 \pm 3.1$	$-1.3 \pm 2.2$	$2.9 \pm 1.1$
<b>Methyl aromatics</b>			
benzene	$59.2 \pm 21.9$	$43.6 \pm 9.6$	$42.8 \pm 13.5$
toluene	$100.0 \pm 4.2$	$100.0 \pm 7.1$	$100.0 \pm 6.6$
styrene	$27.0 \pm 8.2$	$28.7 \pm 8.4$	$43.8 \pm 10.0$
cumene	$21.9 \pm 10.6$	$15.8 \pm 7.0$	$9.2 \pm 5.7$
p-cymene	$-3.7 \pm 4.3$	$-2.3 \pm 3.0$	$-4.0 \pm 6.9$
o-xylene	$-2.3 \pm 2.9$	$0.2 \pm 0.5$	$-2.7 \pm 3.5$
m-xylene	$-1.4 \pm 1.8$	$-0.5 \pm 1.3$	$-0.3 \pm 5.3$
p-xylene	$2.0 \pm 1.8$	$-0.9 \pm 1.9$	$1.9 \pm 0.4$
<b>Side chain oxidation</b>			
benzylalcohol	$7.5 \pm 2.7$	$7.3 \pm 2.9$	$8.8 \pm 4.5$
benzaldehyde	$105.0 \pm 25.5$	$43.8 \pm 11.2$	$34.2 \pm 7.2$
benzoate	$1.3 \pm 3.4$	$-1.2 \pm 2.9$	$5.6 \pm 3.0$
o-toluate	$1.8 \pm 2.4$	$-2.8 \pm 0.7$	$-2.0 \pm 1.5$
m-toluate	$-0.1 \pm 0.9$	$2.7 \pm 2.8$	$-2.0 \pm 2.4$
p-toluate	$2.9 \pm 1.9$	$4.0 \pm 2.2$	$-6.3 \pm 6.8$
<b>Phenols</b>			
phenol	$11.4 \pm 3.5$	$16.0 \pm 9.8$	$14.2 \pm 3.4$
o-cresol	$5.5 \pm 3.9$	$12.6 \pm 5.5$	$10.2 \pm 3.3$
m-cresol	$18.2 \pm 4.5$	$19.9 \pm 9.0$	$19.8 \pm 5.9$
p-cresol	$6.2 \pm 4.5$	$6.3 \pm 5.9$	$12.3 \pm 3.4$
2-fluorophenol	$4.8 \pm 2.4$	$5.1 \pm 4.6$	$5.3 \pm 1.1$
3-fluorophenol	$13.6 \pm 4.6$	$12.0 \pm 3.3$	$10.8 \pm 1.4$
4-fluorophenol	$9.3 \pm 4.1$	$18.7 \pm 4.4$	$19.0 \pm 5.9$
2-chlorophenol	$0.8 \pm 2.3$	$2.7 \pm 2.4$	$3.1 \pm 0.8$
3-chlorophenol	$15.5 \pm 4.3$	$16.8 \pm 5.3$	$18.6 \pm 6.9$
4-chlorophenol	$14.1 \pm 3.1$	$12.1 \pm 4.5$	$20.1 \pm 4.0$
<b>Others</b>			
glucose	$12.8 \pm 4.6$		$-1.7 \pm 0.8$

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1 Table 3: Specific oxygen uptake activity of the (chloro)catechol-1,2-dioxygenase of strain FLU100  
 2 after growth on fluorobenzene (FB), toluene (T) or benzene (B). The value in parenthesis  
 3 represents the number of independent batch cultures tested twice. The first value  
 4 describes the average oxygen uptake while the second one describes the variation. The  
 5 absolute activity of fluorobenzene grown cells was  $2090.5 \pm 395.1$  units,  $189.4 \pm 124.5$   
 6 units for toluene and  $88.7 \pm 29.0$  units for benzene. One unit of oxygenase activity was  
 7 defined as the conversion of  $1 \mu\text{g O}_2 \cdot \text{L}^{-1} \cdot \text{min}^{-1} \cdot \text{OD}^{-1}$ .  
 8 B: benzene T: toluene FB: fluorobenzene

	Specific activity FB with reference T (7)	Specific activity FB with reference catechol (7)	Specific activity T with reference T (7)	Specific activity T with reference catechol (7)	Specific activity B with reference T (7)	Specific activity B with reference catechol (7)
<b>Catechols</b>						
catechol	$465.7 \pm 128.0$	$100.0 \pm 5.1$	$30.5 \pm 10.5$	$100.0 \pm 9.6$	$16.0 \pm 5.5$	$100 \pm 12.3$
3-methylcatechol	$598.6 \pm 128.6$	$136.1 \pm 16.3$	$43.2 \pm 14.6$	$145.2 \pm 32.9$	$80.5 \pm 25.4$	$571.6 \pm 247.2$
4-methylcatechol	$504.8 \pm 87.9$	$119.8 \pm 25.6$	$20.9 \pm 13.4$	$68.1 \pm 34.2$	$33.8 \pm 8.2$	$232.9 \pm 91.1$
3-methoxycatechol	$400.4 \pm 34.5$	$108.7 \pm 29.7$	$17.8 \pm 9.3$	$50.0 \pm 23.2$	$13.5 \pm 3.7$	$91.2 \pm 36.4$
3-fluorocatechol	$123.4 \pm 11.5$	$33.3 \pm 8.6$	$4.2 \pm 5.5$	$8.4 \pm 13.8$	$1.5 \pm 3.4$	$8.6 \pm 21.4$
4-fluorocatechol	$376.4 \pm 99.7$	$83.8 \pm 4.3$	$28.1 \pm 7.0$	$97.0 \pm 43.9$	$31.8 \pm 12.7$	$195.5 \pm 51.1$
3-chlorocatechol	$275.8 \pm 32.0$	$65.9 \pm 18.3$	$19.7 \pm 10.2$	$63.7 \pm 20.5$	$33.5 \pm 9.7$	$235.5 \pm 83.1$
4-chlorocatechol	$95.1 \pm 11.2$	$23.9 \pm 10.2$	$13.0 \pm 5.6$	$43.2 \pm 20.1$	$23.4 \pm 9.9$	$151.0 \pm 72.9$
2,3-dihydroxybenzoate	$-3.3 \pm 0.7$	$-0.9 \pm 0.2$	$-0.7 \pm 1.9$	$-2.7 \pm 6.5$	$-1.6 \pm 1.5$	$-7.6 \pm 6.8$
3,4-dihydroxybenzoate	$9.4 \pm 2.3$	$3.1 \pm 1.4$	$11.7 \pm 6.3$	$38.0 \pm 22.4$	$16.4 \pm 5.4$	$125.8 \pm 74.7$

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10 Table 4: Identification of 3-methylcatechol (native form) and 2-methoxy-3-methylphenol  
 11 (methylated form) as intermediate of toluene degradation of strain FLU100 by GC-MS  
 12 analyses. The fragment with the highest intensity is normalised to 100 % and other  
 13 fragments are given as relative intensities.

Identity	Fragment [m/z]	Intensity [%]	Description
3-methylcatechol	124	100.0	M <sup>+</sup>
	123	35.0	M <sup>+</sup> - H
	105	7.4	M <sup>+</sup> - H - H <sub>2</sub> O
	95	4.0	M <sup>+</sup> - HCO
	78	24.0	M <sup>+</sup> - HCO - OH
	77	11.2	M <sup>+</sup> - CH <sub>3</sub> - OH - OH - H
	2-methoxy-3-methylphenol	138	100.0
123		94.2	M <sup>+</sup> - CH <sub>3</sub>
95		6.8	M <sup>+</sup> - CH <sub>3</sub> - HCO
77		12.7	M <sup>+</sup> - OCH <sub>3</sub> - OH - CH

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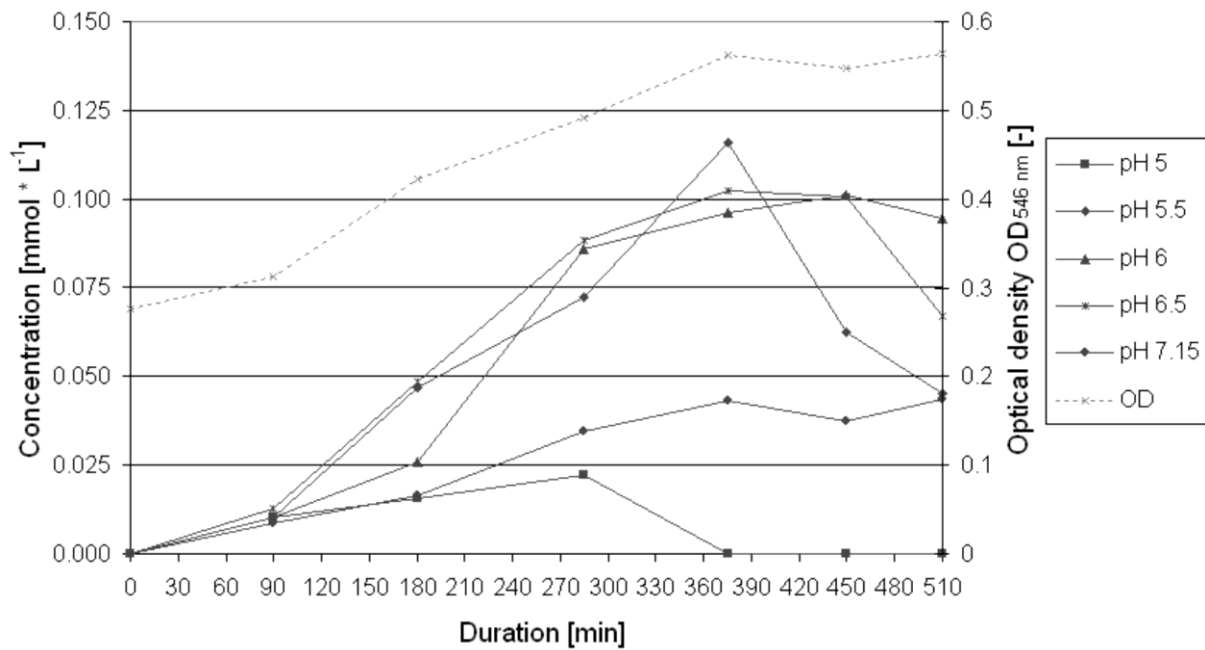
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1 Table 5: Conversion rates (in  $\text{mmol}\cdot\text{L}^{-1}\cdot\text{h}^{-1}\cdot\text{OD}_{546}^{-1}$ ) and generation times (in h) of cells of FLU100  
 2 pre-grown on toluene or glucose as reference.  
 3 -: not analysed n.d. not detectable

	Toluene without CAP	with CAP	generation time [h]	Glucose without CAP	with CAP	Generation time [h]
3-methylcatechol	0.18	-	8.6	-	-	-
acetate	0.80	n. d.	9.5	0.69	n. d.	19.1
pyruvate	1.48	n. d.	10.1	-	-	-
succinate	3.33	n. d.	2.7	-	-	-
methyl succinate	n. d.	n. d.	30.9	-	-	-
adipate	0.62	0.01	4.5	-	-	-
$\beta$ -hydroxy butyrate	0.79	n. d.	7.3	0.47	0.11	13.6
$\gamma$ -hydroxy butyrate	n. d.	n. d.	51.2	0.03	0.03	> 100
2-methyl butyrate	n. d.	n. d.	> 100	n. d.	n. d.	98.6
$\gamma$ -hydroxy butyrolactone	n. d.	0.06	> 100	n. d.	0.01	> 100
malonate	n. d.	n. d.	> 100	n. d.	n. d.	> 100
3-oxoglutarate	0.37	0.48	10.5	0.18	0.04	5.1
2-ML	0.41	0.04	5.4	-	-	-

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6 Fig 1: Influence of the pH value of the medium on the concentration of 2-methylmuconic acid as  
 7 intermediate during conversion of  $1.55 \text{ mmol}\cdot\text{L}^{-1}$  by strain FLU100 pre-grown on  
 8 toluene. The chart for pH 5.0 fell out of the series due to reduced conversion rates of  
 9 toluene as substrate.  
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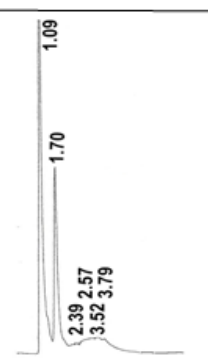
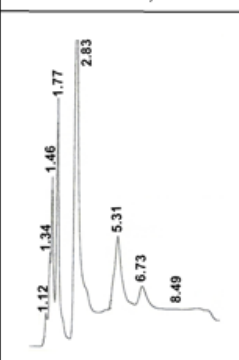
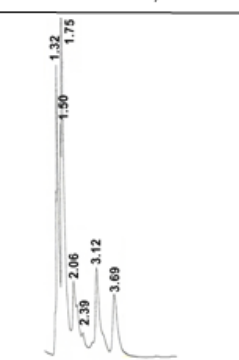
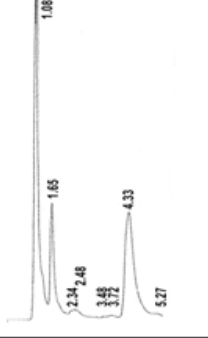
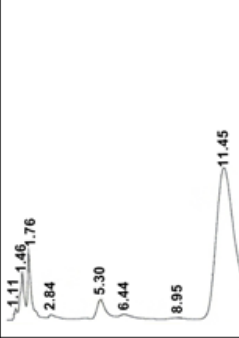
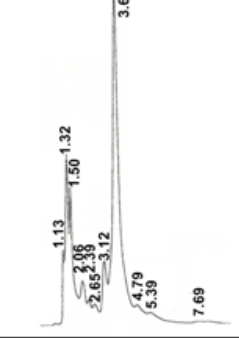

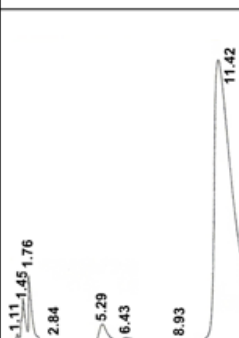
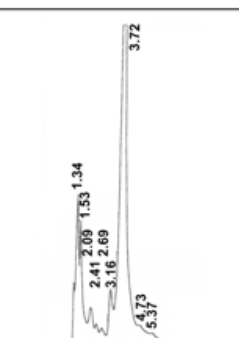
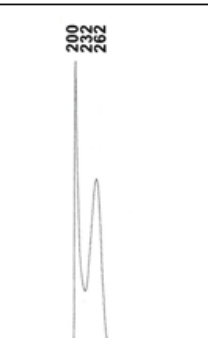


substrate	benzene	toluene	fluorobenzene
Corresponding ,diendiol'	cyclohexa-3,5-diene-1,2-diol	3-methylcyclohexa-3,5-diene-1,2-diol	3-fluorocyclohexa-3,5-diene-1,2-diol
HPLC chromatogram with corresponding ,diendiol'			
HPLC chromatogram after acidification (pH 1-2) and denaturation (80°C, 20')			
spiking substance	phenol	2-methylphenol	2-fluorophenol
HPLC chromatogram after spiking			
Wavelength spectrum ,diendiol'			
Retention time ,diendiol'	1.65 min	2.83 min	1.75 min

Fig 2: Charakterisation and identification of formed 'diendiol' structures of benzene, toluene and fluorobenzene by mutant strain FLU100 P2R5 via HPLC analyses (column: ProntoSIL™ SC-04 Eurobond C18 column, 125 mm · 4 mm, i.d. 5 µm; solvent: H<sub>2</sub>O : CH<sub>3</sub>OH : H<sub>3</sub>PO<sub>4</sub> (85 w/v%) = 74.9 % : 25 % : 0.1 %). The flow rate was maintained at 1 mL · min<sup>-1</sup>.

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