

# Development, validation, and application of an acetylcholinesterase-biosensor test for the direct detection of insecticide residues in infant food

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**Abstract**

A highly sensitive and rapid food-screening test based on disposable screen-printed biosensors was developed, which is suitable for monitoring infant food. The exposure of infants and children to neurotoxic organophosphates and carbamates is of particular concern because of their higher susceptibility to adverse effects. The European Union has therefore set a very low limit for pesticides in infant food which must not contain concentrations exceeding 10 µg/kg for any given pesticide. The maximum residue limit (MRL) has been set to be near the determination threshold that is typically achieved for pesticides with traditional analytical methods. The biosensor method could detect levels lower than 5 µg/kg and thus clearly fulfills the demands of the EU. To substantiate these measurements, recovery rates were determined and amounted on average to 104 % in food. Matrix effects were eliminated by the introduction of a special electrode treatment. The test was compared with two traditional pesticide multiresidue analysis methods (GC/MS, LC/MS) using 26 fruit and vegetable samples from local markets and 23 samples of processed infant food from Germany, Spain, Poland and the USA. Three infant food samples exceeded the MRL of 10 µg/kg when analyzed by either biosensor test or multiresidue methods.

*Keywords:* acetylcholinesterase biosensor, organophosphate, carbamate, validation, food, infant food

## 1. Introduction

Pesticides are widely used in agriculture to increase yield, control microorganisms, which might produce toxic or carcinogenic metabolites, and reduce the costs of food production. According to statistical data published by the Food and Agriculture Organization of the United Nations (FAO), more than 1.2 million metric tons of pesticides were sold to the agricultural sector world-wide in 1995 (FAOSTAT, 1995). These pesticides could be divided into 26 % insecticides, 31 % fungicides and 43 % herbicides. The majority of the insecticides used were acetylcholinesterase inhibitors, 55 % of them belonged to the group of organophosphates and 11 % to carbamates, whereas the others were pyrethroids, chlorinated hydrocarbons or other insecticides. These pesticides are intended to kill living organisms, and a potential dose-related acute and chronic toxicity exists in humans (Jeyaratnam, 1990). Of particular concern is the exposure of infants and children to food contaminants because of their possible increased susceptibility to adverse effects (Larsen and Pascal, 1998; Koletzko et al., 1999). The toxicity of pesticides in infants and children may differ quantitatively and qualitatively from that in adults (Dencker and Ericsson, 1998; Ostergard and Knudsen, 1998; Schilter and Huggett, 1998; Karalliedde, 1999).

Considering the multitude of risks associated with pesticide intake by infants, the European Union has set a very low limit for pesticides in infant food. According to this regulation infant formulae must not contain residues of individual pesticides at levels exceeding 10 µg/kg (EC, 1999). This threshold has been set close to the value that can be analytically determined with traditional methods. In the USA, the 1996 Food Quality Protection Act mandated increased attention to the potential risks to infants and children from pesticide residues which resulted in a 10-fold safety factor for all pesticides to which children are exposed, unless the Environmental Protection

Agency (EPA) can show that an adequate margin of safety is assured without it (FQPA, 1996; Scheuplein, 2000).

Consequently, there is a growing interest in fast and more sensitive detection systems. Traditional methods for insecticide detection are based on gas chromatography (GC) or high performance liquid chromatography (HPLC) coupled with mass selective detectors (MSD) (Martinez et al., 1992; Pylypiw, 1993). Depending on the food being investigated, e.g. citrus fruits, different multiresidue methods for the compound-specific detection of pesticides are available (Anastassiades and Scherbaum, 1997). The disadvantage of these methods is the relatively long assay time and their inherent restriction to a limited number of pesticides, which can be identified by the multiresidue methods. Hence, the possibility of detecting false negatives in unknown samples exists if not all pesticides are covered by the applied multiresidue method. As an alternative, acetylcholinesterase (AChE) inhibition tests, and the AChE-biosensors in particular, have been repeatedly described for use in insecticide detection (Guilbault and Ngeh-Ngwainbi, 1988; Palleschi et al., 1992; Skladal and Mascini, 1992; Mionetto et al., 1994; Saul et al., 1995; Evtugyn et al., 1996; Ghindilis et al., 1996; Makower et al., 1997; Abad et al., 1998; Bachmann and Schmid, 1999; Bachmann et al., 2000). As these tests can be used to determine the presence of organophosphates or carbamates as sum parameter of AChE inhibition with a high level of sensitivity in a relatively short time, AChE biosensors can prove useful in monitoring infant food, which must be free of pesticides and, in particular, free of neurotoxic compounds.

Most of the AChE biosensors described so far are limited to analyzes based on aqueous solutions because of their possible susceptibility towards matrix effects. The few reports on cholinesterase-based biosensors applied to food testing either rely on laborious multi-step sample preparations or prove problematic because of the matrix

effects which reduce the accuracy of the results (Saul et al., 1995; Palchetti et al., 1997; Skladal et al., 1997; Nunes et al., 1998; Nunes et al., 1999; Pogacnik and Franko, 1999). A recently developed test applying direct incubation in solvent extracts of food samples performed well in food with a high water content but showed reduced recovery rates of paraoxon in food with a low water content (Schulze et al., 2002).

In this paper, we describe an amperometric biosensor test based on screen-printed AChE biosensors which were described by Bachmann and Schmid (Bachmann and Schmid, 1999). The biosensor can be applied directly in food samples without the laborious pre-treatment of the samples. Matrix effects could be eliminated with special electrode treatment. In order to validate the performance of the method it was compared with traditional analytical methods used at the Chemical and Veterinary Official Laboratory (CVUA) Stuttgart ("Chemisches und Veterinär-Untersuchungsamt Stuttgart"), Germany. One of the methods applied involves classical solvent-based extraction and partitioning. The other method involves supercritical fluid extraction. Determination was performed by GC/MS and LC/MS. The developed biosensor method proved sensitive enough to be applied in the analysis of infant food samples.

## **2. Experimental**

### *2.1. Reagents and samples*

AChE (EC 3.1.1.7) from electric eel (Type V-S, 970 U/mg) was obtained from Sigma-Aldrich (Deisenhofen, Germany). Paraoxon (paraoxon-ethyl) and Tween-20 were purchased from Riedel de Haën (Seelze, Germany). Pesticide standards were purchased from Promochem (Wesel, Germany) or Riedel de Haën (Seelze, Germany). Insecticide stock solutions were prepared in ethanol. All other reagents

were of analytical grade as supplied by Sigma-Aldrich (Deisenhofen, Germany). Hydromatrix<sup>®</sup> no. 0019-8004 for SFE was obtained from Varian (Darmstadt, Germany). Acetonitrile, methanol, cyclohexane, isooctane, ethyl acetate and acetone were all of residue analysis grade.

Infant food samples were obtained from local stores in Germany, Spain, Poland and the USA. Five different German brands, one Polish brand (in four different flavors), two Spanish brands and two American brands were tested.

### *2.2. Expression of human AChE synthetic gene in Pichia pastoris*

The following strategy (in brief) was used for cloning and expression of human AChE (huAChE) (Vorlová, S., Schmitt, J. and Schmid, R.D., in preparation): A synthetic gene of the huAChE was completely synthesized following the method of mutually priming long overlapping oligonucleotides. The nucleotide sequence was optimized for the expression in the yeast *P. pastoris*. The gene was inserted into the shuttle vector pPICZ $\alpha$ A under the control of the *AOX1* promoter; transformation of yeast cells was performed by electroporation. Recombinant *P. pastoris* transformants producing acetylcholinesterase were grown according to the manufacturer's (Invitrogen, Carlsbad, CA, USA) protocol in glycerol-containing (1 % (v/v)) BMGY media and induced in BMMY media, containing 0.5 % (v/v) methanol in agitated (200 rpm) shaking flasks. Every 24h, 0.5 % (v/v) of pure methanol was added to maintain induction, and an aliquot of the culture was withdrawn to monitor the expression level of huAChE in the cells and the culture supernatant.

### *2.3. Biosensor measurement*

Disposable biosensors with immobilized electric eel and human AChE were produced following the screen-printing method suggested by Bachmann and Schmid (Bachmann and Schmid, 1999). In brief, thickfilm electrodes were printed on flexible

polyvinylchloride sheets from SKK (Denzlingen, Germany) using a DEK 249 screen printer (DEK Ltd., Weymouth, England) and polyester screens (45°, 100 T) that were purchased from Steinmann GmbH (Stuttgart, Germany). Screen printing inks (Electrodag PF-410, Electrodag 423 SS, Electrodag 6037 SS) were obtained from Acheson (Scheemda, Netherlands). Ink for working electrodes contained 3% (w/w) hydroxyethyl cellulose (HEC) and 15% (w/w) 7,7,8,8-tetracyanoquino-dimethane (TCNQ)-graphite (2,5% TCNQ in T15 graphite from Lonza (Basel, Switzerland)). Marastar SR 057 purchased from Marabu (Tamm, Germany) was used as insulation ink. The electrodes were cured for 30 min at 90°C before enzyme immobilization. The enzyme printing ink contained 5 % (w/w) bovine serum albumine (Sigma, Deisenhofen Germany) and 1 U/mL human or electric eel AChE in 1 % (w/w) hydroxyethyl cellulose (Fluka, Buchs, Switzerland) aqueous solution. This solution was printed with the screen-printing device on top of the working electrodes. Immobilization by cross-linking was performed in glutaraldehyde vapor for 15 min at room temperature. Human AChE was directly used from the culture supernatant without further purification steps. All sensor experiments were carried out in a stirred buffer solution (0,01 M potassium phosphate buffer, 0,05 M NaCl, pH 7,5 (PBS)) at room temperature. Enzyme activity was determined by monitoring thiocholine formed by enzymatic hydrolysis of acetylthiocholine chloride (1 mM). Thiocholine was determined by oxidation at 100 mV versus Ag/AgCl. For use in the inhibition experiments, the biosensor was incubated with a sample for 30 min at room temperature in a non-stirred solution and percentage of inhibition was calculated after the measurement of residual activity.

#### *2.4. Reactivation of AChE-activity*

To determine the AChE-activity reactivation rate, 1 mM pyridine-2-aldoxime methochloride (2-PAM) in PBS was used as reactivating agent. The biosensor was incubated for 30 min in a stirred reactivation solution at room temperature. The final AChE activity was then measured in a buffer solution.

#### *2.5. Biosensor food test*

Food samples (10 g) were mixed with 10 mL 1 M potassium phosphate buffer (PBS) (pH 7,5) to ensure a sufficient amount of water at a neutral pH value. The AChE biosensor was incubated in the food sample for 30 min. Before measuring the activity of the enzyme in the buffer solution, the electrodes were placed in PBS containing 1 vol-% Tween-20 for 15 min and then washed with pure PBS before measuring the initial AChE activity and after incubation in the food sample before measuring the residual AChE activity.

#### *2.6. Conventional analytical methods*

##### *2.6.1. Supercritical fluid extraction (SFE) method*

A representative sample portion of 7.5 g was weighed into a 100 mL beaker. Six g Hydromatrix<sup>®</sup> (Varian, no. 0019-8004) was added and the mixture stirred intensively to achieve good homogeneity. An aliquot of 4.5 g (2.5 g sample) was transferred into the extraction thimble. To prevent particles from being extruded during extraction or depressurization, which could result in tube plugging, a fine grade glass wool paper was placed at either end of the extraction thimble.

The loaded thimbles were placed into a freezer immediately after filling and kept there until further processing because previous storage experiments had shown that some analytes (e.g. ethiofencarb, methiocarb, benfuracarb, furathiocarb, carbosulfan,

dichlorvos) degraded significantly in the extraction cells when kept at room temperature (Anastassiades, in preparation). Freezing of the loaded thimbles (at  $-18^{\circ}\text{C}$ ) efficiently prevented degradation.

A second 4.5 g aliquot of the sample/Hydromatrix<sup>®</sup> mixture was stored in a glass vial for further use in case the extraction had to be repeated.

Extraction of the samples by SC-CO<sub>2</sub> and reconstitution of the extracts in acetonitrile was performed automatically. The extraction conditions used are shown in Table 1.

### *2.6.2. Liquid solvent based method*

A modified version of a multiresidue method was used to extract organophosphates and carbamates, which was developed at the CVUA Stuttgart for the analysis of pesticide residues in citrus fruits (Anastassiades and Scherbaum, 1997). After sample comminution, 20 g sample were extracted with 50 mL acetone with an Ultra-Turrax and filtrated. Water and 10 mL saturated NaCl solution was added to the extract. Two liquid-liquid partitioning steps were performed with 20 mL cyclohexane/ethyl acetate (1:1). The combined extracts were dried with Na<sub>2</sub>SO<sub>4</sub> and concentrated, followed by reconstitution with acetonitrile to a defined volume of 10 mL. Determination was performed with GC in SIM mode or LC/MS in ESI-positive mode.

### *2.6.3. Measurement*

The instrumental conditions used during LC/MS in ESI (pos.) mode for the identification and quantification of the carbamates are listed in Table 2. An internal standard calibration was performed for quantification. Table 3 shows the target and qualifier ions used. Determination limits for the carbamates were 0.01 mg/kg or lower. Organophosphates were analyzed using GC-MS in single ion monitoring (SIM) mode in order to achieve good determination limits (HP 5890 Series II, auto sampler 7673,

injector Gerstel CIS, solvent mode, detector HP 5970 MSD, equipped with uncoated pre-column 3 m x 0.25 mm, press fit connection to the analytical column HP5MS, 30 m x 0.25 mm, 0.25  $\mu$ m, carrier gas helium constant flow). Table 4 shows the spectrum of the compounds and the target and qualifier ions used for SIM mode. The determination limit was 0.01 mg/kg or lower for most compounds (see also Table 4).

### **3. Results and discussion**

#### *3.1. Preventing matrix effects with electrode modification*

To develop a simple assay system, it was attempted to measure untreated food with the AChE electrode directly. Aliquots of orange juice, peach and apple infant food were used for this purpose and a high AChE inhibition was observed in all cases (Table 5). An expected reason for the high inhibition rate was the low natural pH value of fruits: adjustment to a neutral pH reduced the inhibition rate from 56 % to 41 % in the case of orange juice. AChE inhibition could not be reactivated with 2-PAM, a nucleophilic oxime. Oxime treatment removes the phosphate group from the serine residue of the inhibited AChE. It can therefore be concluded that the AChE inhibition did not result from the presence of organophosphates or carbamates in the food samples. In addition, it was observed that during activity measurements the thickfilm electrodes required much longer equilibration times when they were previously incubated in food samples. The equilibration time denotes the time that is required until a constant current is reached and the substrate can be added to the buffer solution. We believe that these problems are caused by matrix particles that are bound to the surface of the electrodes. These matrix particles could not be removed from the electrode surface by washing it with either buffer or water. One possibility to prevent these matrix effects is to add an extraction step in the protocol followed by

incubation in the solvent extract, as described by Schulze and coworkers (Schulze et al., 2002). In the present study, matrix effects were avoided by treating the electrodes with a 1 vol-% Tween-20 solution in phosphate-buffer (pH 7,5) after their incubation in the food samples. The use of this detergent solution solved virtually all the previously described problems. Unspecific inhibition was no longer observed and the time required for equilibration took approximately 10 minutes. The detergent solution (1 vol-% Tween-20) itself did not lead to a decrease in AChE activity.

### *3.2. Test performance*

The proper performance of the test was verified by analyzing several food samples to which different amounts of paraoxon-ethyl in the concentration range between 5 and 20 µg/kg were added. The inhibition of electric eel AChE (eeAChE), caused by the incubation in the spiked food, was compared with the inhibition that was observed when equivalent paraoxon concentrations were present in the buffer-solution. Recovery rates were calculated out of the difference of these two inhibition values. Figure 1 shows the AChE inhibition values of spiked peach-based infant food. The inhibition values correlated well with the standard inhibition curve in buffer solution. The recovery rate for paraoxon-ethyl in peach infant food was 101 %. Experiments with apple-based infant food yielded a recovery rate of 106 % (Figs. 2). All recovery rates met the legal requirements of a value between 70 and 110 % (EC, 1997). The analysis of spiked food samples showed an acceptable reproducibility even at very low inhibitor concentrations. The coefficient of variation was on average 8 % and 9 % in the case of peach and apple infant food.

The determination limit in food was less than 5 µg/kg paraoxon with eeAChE. The signal- to-noise ratio at a concentration of 5 µg/kg paraoxon ( $1,8 \times 10^{-8}$  mol/L) in apple and peach-based infant food was well above 3 : 1 (e.g. 8 : 1 in the case of

peach-based infant food). Lower paraoxon concentrations were not tested in food samples. The determination limit was similar to that measured with conventional analytical methods and could be reduced using AChEs from other sources, such as e.g. *Drosophila melanogaster* AChE or genetically engineered AChE variants (Villatte et al., 1998; Schulze et al., in press).

With respect to pesticides the developed test has several advantages over standard analytical methods and previously described biosensor tests. Using this test, samples can be analyzed within approximately 90 min, which means that at least four to five hours can be saved with the test compared to the commonly used multiresidue methods (Anastassiades and Scherbaum, 1997). The handling of the samples is much easier than with traditional methods since the test does not involve complicated extraction, cleanup and measurement steps which often require well-trained personnel and sophisticated equipment. This test also avoids the use of any organic solvent and involves no pre-concentration step. In contrast to other biosensor tests, which showed reduced recovery rates in food (in most cases below 70 %) (Skladal et al., 1997; Nunes et al., 1998; Nunes et al., 1999; Pogacnik and Franko, 1999), the accuracy of the recoveries is very high, even at very low neurotoxin concentrations.

The main advantage of the biosensor tests over conventional methods is that principally all AChE-inhibiting compounds can be detected. However, it is only possible to measure a sum parameter of AChE inhibition; qualitative or quantitative information about the single compounds cannot be obtained. In combination with traditional multiresidue methods, the biosensor test can nevertheless be applied as a pre-screening test to discriminate between samples that do or do not contain AChE inhibitors. Should AChE inhibitors be targeted, the biosensor test would help to increase the number of tested samples. To minimize false positive results, it is advisable to set an inhibition threshold for each commodity (action level).

### 3.3. Validation with standard analytical methods

To validate the biosensor method, 26 fruit and vegetable samples were tested which were previously analyzed with traditional multiresidue methods. Two types of AChEs were used for these measurements: electric eel and recombinant human AChE. Ratios of AChE inhibition of these two enzymes were compared with ratios of bimolecular rate constants of AChE inhibition ( $k_i$ ). The results are shown in Table 6. Only a single insecticide could be detected in the first five samples with the multiresidue method. A pear sample, which contained 22  $\mu\text{g}/\text{kg}$  carbofuran, led to 48 % eeAChE inhibition and to 25 % huAChE inhibition. The different extent of enzyme inhibition results from the differences in the  $k_i$  values of eeAChE and huAChE towards these neurotoxins. The low inhibition observed with the apple sample containing 158  $\mu\text{g}/\text{kg}$  pirimicarb was expected because of the small  $k_i$ -values of  $1,0 \cdot 10^3 \text{ M}^{-1} \text{ min}^{-1}$  and  $1,0 \cdot 10^4 \text{ M}^{-1} \text{ min}^{-1}$ , respectively. The ratios of enzyme inhibition of the first four samples (1,9 / 0,8 / 0,3 / 0,9) corresponded with the expected values calculated from the  $k_i$  values (3,6 / 1,1 / 0,1 / 0,8). To the best of our knowledge,  $k_i$ -values have so far not been published for methamidophos.

The next five samples (Table 6) contained a mixture of different neurotoxins. Pears were contaminated with 50  $\mu\text{g}/\text{kg}$  carbaryl, 17  $\mu\text{g}/\text{kg}$  chlorpyrifos ethyl and 20  $\mu\text{g}/\text{kg}$  chlorpyrifos methyl. This pear sample caused a 71 % inhibition of eeAChE. The reason for the observed inhibition caused by the next three samples (lemon, orange and grapefruit) could not finally be clarified. One possible explanation is that small amounts of the present phosphorothionates which do not inhibit AChE at this concentrations are spontaneously oxidized to the corresponding oxo-variants, e.g. by the impact of UV light (Chambers and Levi, 1992). Finally, no pesticides could be detected with the GC-MS method in a number of samples. Most of these did not lead to any or only slight inhibition of the applied acetylcholinesterases. Others, however,

inhibited the enzymes significantly (more than 10 %). A possible explanation for this observation is the limited analyte spectrum of the GC-MS analysis. The multiresidue methods usually used to examine fruits and vegetables cover just 200 of about 500 applied pesticides and many AChE inhibitors are not included.

The correct functioning of the biosensor test could be confirmed in many different types of fruits and vegetables, from the very fibrous pea or asparagus to several kinds of citrus fruits. The use of a multi-electrode system employing two different AChEs, which usually have different inhibition constants towards the tested organophosphates and carbamates, gives additional certainty about the observation that the inhibition is caused by neurotoxins and not by some kinds of matrix effects. The use of human AChE can also help to estimate the potential health risk of the analyzed food to man.

#### *3.4. Infant food analysis*

Processed infant food samples from Germany, Spain, Poland and the USA were analyzed with the AChE-biosensor test and with the multiresidue method involving LC-MS and GC-MS analysis. In total, 23 different types of infant food were tested: 13 different flavors of four German brands, four flavors of two Spanish brands, four flavors of one Polish brand and two flavors of two American brands. The biosensor measurements were again performed with a two-electrode system employing electric eel and human AChE. Since the concentrations of possible contaminants in infant food are typically very low, an additional proof of evidence for the functionality of the biosensor was added to the analytical procedure. The determination of the reactivation rate of the inhibited enzyme after reaction with a nucleophilic oxime which reconstitutes the active AChE turned out to be an appropriate way to discriminate between infant food contaminated with neurotoxins and infant food free

of cholinesterase-inhibitors. But reactivation of inhibited AChE is only possible before "aging" (loss of an alkyl residue) of the phosphoryl moiety attached to the active site serine of AChE. This aging process depends on the kind of organophosphate which reacts with AChE, e.g. methoxy groups age more rapidly than ethoxy groups (Chambers and Levi, 1992).

Of 23 tested infant food samples, 13 caused AChE inhibition, which could be at least partially reactivated (see Table 7). The multiresidue method detected pesticides in 11 of the samples tested. Two samples led to reactivatable AChE inhibition but no neurotoxins were detected by LC-MS or GC-MS. In this case, the restriction of the standard detection methods to a limited number of pesticides must be taken into account. The other 10 tested infant food samples did not show any reactivatable AChE inhibition. These samples were not analyzed with the multiresidue methods. Of the 14 detected pesticides, the following proved to be AChE inhibitors: pirimicarb, ethiofencarb, carbaryl, methiocarb and paraoxon. Carbaryl was the most commonly found pesticide.

Three violations of the maximum residue limit of 10 µg/kg valid in the EU for infant food were observed. One batch of the German banana-tangerine-pear infant food contained 20 µg/kg pirimicarb and 50 µg/kg ethiofencarb. The analysis of this sample with huAChE resulted in 10 % inhibition, which could be reactivated to an extent of 68 %. In samples of an older batch of the same brand no pesticides were detected by either of the two methods. A German fruit salad infant food contained 10 µg/kg carbaryl. Seventy µg/kg propiconazol and 20 µg/kg esfenvalerat were found in an American peach infant food sample. These two compounds are no cholinesterase inhibitors, but it is not unlikely that also pesticides, in addition to those detected, might be present in the sample.

Concerning the detection frequency of pesticides in infant food (EWG, 1995), a study carried out by the Environmental Working Group (EWG), an American non-profit environmental research organization, came to similar results. Of 72 infant food samples from the three largest American infant food producers, more than half (53 %) contained detectable levels of pesticides. Two or more pesticides were found in 18 % of the samples. Among the 16 detected pesticides were 9 organophosphates or carbamates. A test of 155 infant food samples conducted with the commercially available Charm Pesticide Test, which is also based on cholinesterase-inhibition, resulted in 14 % positive samples (Saul et al., 1995). Carbaryl was also the most frequently found pesticide in this study. This test has the disadvantage that it requires a laborious sample pre-treatment protocol including the extraction with organic solvents and subsequent re-dissolving in water.

The applicability of the new biosensor test for monitoring infant food could be shown in the present study. Good correlation was observed between the results obtained with the biosensor test and those with the GC-MS and LC-MS. The biosensor test has the advantage over standard analytical methods that the risk of false negative results of the particularly dangerous group of organophosphates and carbamates is minimized substantially. In contrast to the Charm Pesticide Test it requires less working steps and does not require the use of organic solvents.

#### **4. Conclusion and outlook**

A biosensor test for the direct detection of organophosphates and carbamates in food was developed which requires no extraction or pre-concentration steps. Matrix effects could be avoided by the introduction of an electrode pre-treatment method. The test showed excellent recovery rates with acceptable levels of reproducibility even at trace concentrations. Validation was successfully performed by comparing the test

results of real food samples with the conventional multiresidue methods applied at the CVUA Stuttgart. The applicability was proven by analyzing infant food samples. To increase the analyte spectrum of the biosensor test, a combination of different and more sensitive AChE variants could be applied in a multisensor array. It would also be desirable to add a method for the detection of phosphorothionates in food to the test protocol. Currently work is under way to achieve these requirements.

### **Acknowledgement**

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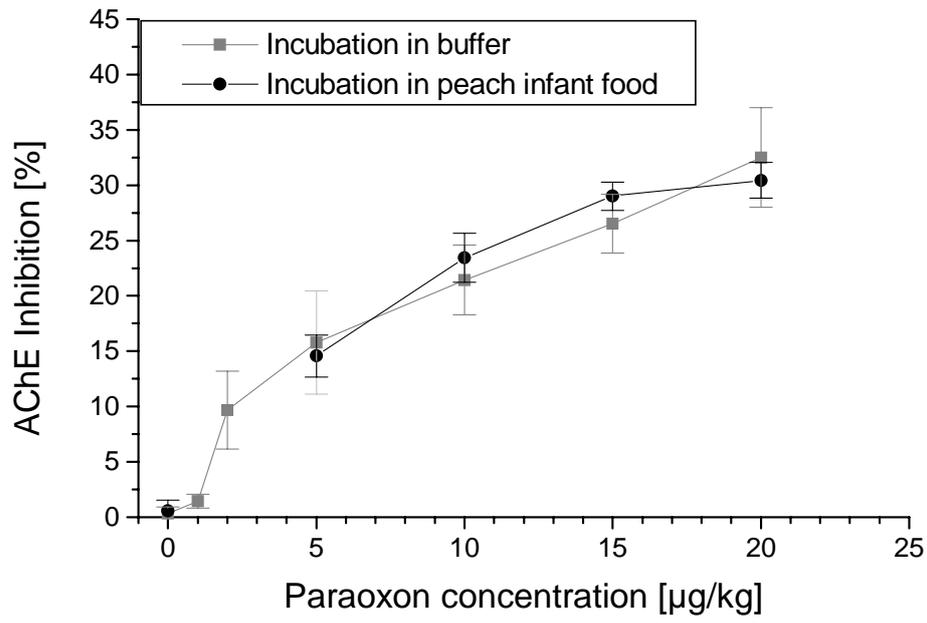


Figure 1:

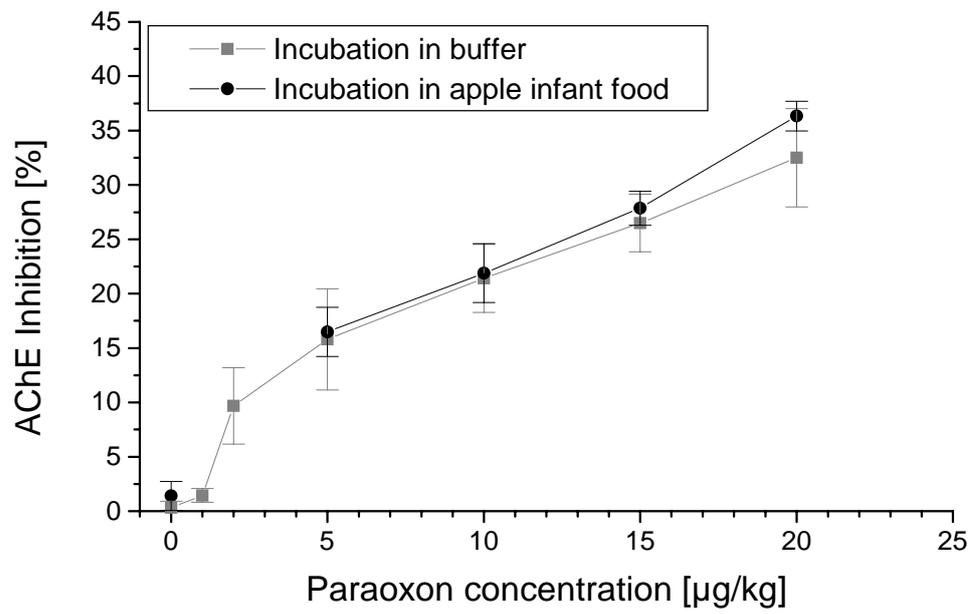


Figure 2:

Table 1: SFE parameters for HP 7680T

Extraction pressure	329 bar
Extraction temperature	55 °C
CO <sub>2</sub> -density	0.89 g/mL
Static extraction time	2 min
Dynamic extraction time	25 min
CO <sub>2</sub> -flow	1.8 mL/min
Trap material	ODS, 10 °C during extraction, 40 °C during elution
Elution solvent	acetonitrile (2 x 1.5 mL, flow 0.5 mL/min)
Solvent for trap clean-up	cyclohexane / ethyl acetate 1:1 (4 mL, flow 0.8 mL/min)
Solvent for trap reconditioning	acetonitrile (3 mL, flow 0.8 mL/min)

Table 2: Instrumental conditions used at LC/MS

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ESI (pos.) mode

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## LC Parameters:

- LC-MSD HP 1100 series
- column: 2.1 x 150 mm, 3.5  $\mu$ , ZORBAX XDB C-18
- injection volume: 5  $\mu$ L
- mobile phases:  
A: 5 mM aqueous  $\text{CH}_3\text{COONH}_4$  : acetonitrile (90:10)  
B: 5 mM aqueous  $\text{CH}_3\text{COONH}_4$  : acetonitrile (10:90)
- flow 0.3 mL/min
- gradient: 30 % B  $\rightarrow$  90 % B in 8 min, temp.: 50  $^\circ\text{C}$

## MSD Parameters:

- API electrospray
  - Variable fragmentor voltage, see Table 3
  - drying gas: nitrogen 10 L/min, 300  $^\circ\text{C}$  Vcap: 4000 V
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Table 3: LC/MS target and qualifier masses and corresponding fragmentor voltages (Frag. [V]) for carbamates

N-Methyl- Carbamate	M+1 (Frag. [V])	M-56 (Frag. [V])	M+18 plus NH <sub>4</sub> <sup>+</sup> (Frag. [V])	M+23 plus Na <sup>+</sup> (Frag. [V])	Other ions (Frag. [V])	Determination limit [mg/kg]
Aldicarb	213 (30)	-	208 (20)		116 (50)	0,005
Aldicarb Sulfon (Aldoxycarb)	223 (65)	166 (100)	240 (30)	245 (100)		0,01
Aldicarb Sulfoxid	207 (40)	-	224 (20)	229 (75)	132 (65) 166 (100)	0,01
Aminocarb	209 (40)	152 (90)				0,005
Bendiocarb	224 (50)	167 (100)	241 (15)		109 (100)	0,01
Carbanolat	214 (50)	157 (90)				0,01
Carbaryl	202 (50)	145 (100)	219 (70)	224 (60)		0,01
Carbofuran	222 (50)	165 (90)			123 (140)	0,005
Dioxacarb	224 (30)	167 (65)			123 (90)	0,01
Ethiofencarb	226 (30)	169 (55)	243 (10)		164 (70) 107 (90)	0,01
Ethiofencarb Sulfon	-	201 (50)			218 (30) 223 (60) 107 (80)	0,01
Ethiofencarb Sulfoxid	-	185 (30)			207 (50) 107 (65)	0,01
Fenobucarb	208 (50)	152 (90)	225 (15)			0,002
Isoprocarb	194 (60)	137 (80)			152 (90)	0,005
Landrin (2,4,5- Trimethacarb)	194 (50)	137 (100)	211 (20)			0,002

N-Methyl- Carbamate	M+1 (Frag. [V])	M-56 (Frag. [V])	M+18 plus NH <sub>4</sub> <sup>+</sup> (Frag. [V])	M+23 plus Na <sup>+</sup> (Frag. [V])	Other ions (Frag. [V])	Determination limit [mg/kg]
Methiocarb	226 (20- 70)	169 (90)			121 (110) 107 (110)	0,005
Methiocarb Sulfon	258 (70)	201 (100)			218 (25) 122 (100) 107 (110)	0,01
Methiocarb Sulfoxid	242 (80- 90)	185 (115)			170 (140) 122 (140) 106 (10)	0,005
Methomyl	163 (40)	106 (80)		185 (80)		0,005
Metolcarb	166 (50)	109 (100)	183 (10)			0,005
Oxamyl	220 (40)	-	237 (10)	242 (80)		0,01
Promecarb	208 (50)	151 (100)	225 (10)		109 (100)	0,002
Propoxur	210 (25)	153 (70)	227 (10)		111 (100) 168 (70)	0,005
Precursor Pesticides						
Thiodicarb	355 (30)	-			193 (60) 149 (70) 108 (90)	0,005
Benfuracarb	411 (50)	-		433 (130)	190 (90) 195 (130)	0,005
Furathiocarb		-		405 (150)	252 (110) 195 (130)	0,005
Carbosulfan	381 (80)	-		419 (90) plus K <sup>+</sup>	160 (120) 118 (130) 195 (130)	0,005

Table 4: List of organophosphates analyzed by GC-MS

Organophosphate	Target and qualifier ions				Determination
					limit [mg/kg]
Acephate	136	94	125		0,01
Azinphos-methyl	160	132	77		0,02
Bromophos	331	329	333	125	0,001
Bromophos-ethyl	359	303	357	301	0,002
Chlorfenvinphos	323	267	269		0,001
Chlorpyrifos	197	199	314		0,001
Chlorpyrifos-methyl	286	288	125	109	0,001
Demeton-S-methyl	88	109	142	230	0,01
Dichlorvos	222	220	185	187	0,01
Dicrotophos	237	192	193		0,01
Dimethoat	125	93	229		0,01
Disulfoton	274	142	186	88	0,002
Ethion	384	231	153		0,001
Ethoprophos	242	200	158	139	0,002
Etrimfos	292	277	181	153	0,001
Fenamiphos	303	288	260	217	0,001
Fenchlorphos	285	125	287	241	0,01
Fenitrothion	277	125	260		0,002
Fenthion	278	125	169		0,002
Fonofos	246	137	109		0,001
Isofenphos	213	255	121		0,002
Malaoxon	268	239	195	127	0,01
Malathion	173	125	127		0,001
Methamidophos	126	94	95	141	0,01
Methidathion	145	85	125		0,005
Mevinphos	192	109	164	127	0,01
Monocrotophos	223	192	193		0,01
Omethoate	156	110	79		0,01
Paraoxon	275	247	220	149	0,01
Paraoxon-methyl	247	230	200	109	0,02
Parathion	291	155	235		0,001
Parathion-methyl	263	125	109	233	0,001
Phorate	260	231	121	75	0,001
Phosalone	367	184	182	121	0,002
Phosmet	160	317	133		0,005
Phosphamidon	264	127	138		0,01

Organophosphate	Target and qualifier ions				Determination
					limit [mg/kg]
Pirimiphos-methyl	290	305	276	233	0,001
Profenofos	208	337	339	139	0,005
Prothiofos	309	267	311		0,001
Pyrazophos	221	232	265	373	0,005
Pyridaphenthion	340	199	188		0,002
Quinalphos	146	298	156		0,002
Tolyfluanid	238	240	181		0,002
Sulfotep	322	266	202	294	0,001
Terbufos	288	231	186	153	0,002
Tolclofos-methyl	265	267	125	250	0,001
Triazophos	257	161	162		0,005
Trichlorfon	221	145	109	147	0,02

Table 5: Effects of untreated food samples at natural and neutral pH values on the activity of electric eel AChE in comparison to effects after Tween-20 treatment (unspecific inhibition):

	AChE Inhibition [%]		
	pH 4	pH 7	Tween-20 pH 7
Apple juice	18	10	3
Orange juice	56	41	2
Peach infant food	21	16	1
Apple infant food	19	12	1

Table 6: Validation of the biosensor test with standard analytical methods

Sample	GC-MS	$I_{ee}$	$I_{hu}$	$I_{ee} / I_{hu}$	$k_{i_{ee}}$		$k_{i_{hu}}$		$k_{i_{ee}} / k_{i_{hu}}$
		[%]	[%]		[M <sup>-1</sup> min <sup>-1</sup> ]		[M <sup>-1</sup> min <sup>-1</sup> ]		
					[1]	[2]	[1]	[2]	
Pear	22 µg/kg Carbofuran	48	25	1,9	$2,9 \cdot 10^6$	$0,8 \cdot 10^6$			3,6
Cherry	40 µg/kg Methiocarb	11	14	0,8	$1,0 \cdot 10^5$	$0,9 \cdot 10^5$			1,1
Apple	158 µg/kg Pirimicarb	1	4	0,3	$1,0 \cdot 10^3$	$1,0 \cdot 10^4$			0,1
Apple	40 µg/kg Dichlorvos	33	38	0,9	$4,2 \cdot 10^4$	$5,2 \cdot 10^4$			0,8
Paprika	110 µg/kg Methamidophos	24	37	0,6					
Apple	114 µg/kg Carbaryl	21	n.a.		$6,6 \cdot 10^4$	$2,7 \cdot 10^4$			2,4
Pear	50 µg/kg Carbaryl	71	n.a.		$6,6 \cdot 10^4$	$2,7 \cdot 10^4$			2,4
	17 µg/kg Chlorpyrifos-ethyl*				$1,8 \cdot 10^6$	$0,6 \cdot 10^7$			0,3
	20 µg/kg Chlorpyrifos-methyl*				$0,7 \cdot 10^6$	$2,2 \cdot 10^6$			0,3
Lemon	123 µg/kg Chlorpyrifos-ethyl*	18	27	0,7	$1,8 \cdot 10^6$	$0,6 \cdot 10^7$			0,3
	813 µg/kg Methidathion*				$0,6 \cdot 10^6$	$0,9 \cdot 10^5$			6,7
Lemon	32 µg/kg Chlorpyrifos-ethyl*	27	31	0,9	$1,8 \cdot 10^6$	$0,6 \cdot 10^7$			0,3
	2470 µg/kg Methidathion*				$0,6 \cdot 10^6$	$0,9 \cdot 10^5$			6,7
Orange	5 µg/kg Chlorpyrifos-ethyl*	21	32	0,7	$1,8 \cdot 10^6$	$0,6 \cdot 10^7$			0,3
	1060 µg/kg Methidathion*				$0,6 \cdot 10^6$	$0,9 \cdot 10^5$			6,7
	838 µg/kg Malathion*				$1,7 \cdot 10^6$	$3,3 \cdot 10^5$			5,2
Grapefruit	180 µg/kg Chlorpyrifos-ethyl*	23	18	1,3	$1,8 \cdot 10^6$	$0,6 \cdot 10^7$			0,3
	830 µg/kg Methidathion*				$0,6 \cdot 10^6$	$0,9 \cdot 10^5$			6,7
Peach	n.d.	13	n.a.						
Apple	n.d.	0	2						
Paprika	n.d.	0	0						
Sugar Pea	n.d.	0	1						
Pea	n.d.	0	1						
Pea	n.d.	1	1						
Salad	n.d.	2	n.a.						
Asparagus	n.d.	3	3						
Asparagus	n.d.	4	7						
Asparagus	n.d.	0	10						
Potato	n.d.	22	9						
Potato	n.d.	5	5						
Gooseberry	n.d.	13	19						
Nisperos	n.d.	5	3						
Plum	n.d.	8	5						

$I_{ee}$ : inhibition of electric eel AChE;  $I_{hu}$ : inhibition of human AChE;  $k_i$ : bimolecular rate constant of AChE inhibition; n.d.: nothing detected; n.a.: not analyzed. \*  $k_i$  values refer to oxidized forms of organophosphates

[1] : P. Herzprung, Methodische Grundlagen des Nachweises und der Bestimmung von insektiziden Phosphorsäureestern und Carbamaten im Wasser mittels Cholinesterasehemmung, *Dissertation TU München* 1991.

[2]: F. Villatte. V. Marcel, S. Estrada-Mondaca, D. Fournier, Engineering sensitive acetylcholinesterase for detection of organophosphate and carbamate insecticides, *Biosensors & Bioelectronics* 13 (1998) 157.

Table 7: Infant food analysis

Infant food sample	I <sub>ee</sub> [%]	R <sub>ee</sub> [%]	I <sub>hu</sub> [%]	R <sub>hu</sub> [%]	GC/LC-MS
Banana-tangerine-pear (Jul01) (Germany)	5	0	10	68	20 µg/kg Pirimicarb 50 µg/kg Ethiofencarb
Banana-tangerine-pear (Jun00)	n.a.		0	0	n.a.
Fruit-salad (Germany)	6	79	9	59	10 µg/kg Carbaryl 6 µg/kg Metalaxyl
Pineapple-banana (Spain)	13	74	5	50	4 µg/kg Metalaxyl
Peach-maracuja (Germany)	7	97	2	100	n.d.
Peach-mango (Poland)	20	20	8	20	1 µg/kg Carbaryl
Carrot (Poland)	8	59	8	50	n.d.
Mixed fruits (Spain)	15	29	16	30	4 µg/kg Methiocarb 4 µg/kg Diphenylamin
Banana-peach-apple (Germany)	11	19	9	25	2 µg/kg Carbaryl
Apricot-yoghurt (Germany)	12	47	6	33	2 µg/kg Methiocarb 2 µg/kg Myclobutanil 1 µg/kg Piperonylbutoxid 3 µg/kg Iprodion
Apricot-pear (Germany)	15	30	6	40	3 µg/kg Methiocarb
Apple-peach (Poland)	30	2	5	0	n.d.
Peaches (USA)	12	21	13	13	70 µg/kg Propiconazol 20 µg/kg Esfenvalerat
Banana (USA)	27	50	29	26	1 µg/kg Carbaryl
Apple-Banana (Germany)	9	0	14	0	n.a.
Fruit-dessert (Germany)	6	3	12	0	n.a.
Banana-peach-apple (Germany)	8	3	3	0	n.a.
Apple (Spain)	13	0	1	0	n.a.
Williams-Christ-Pear (Germany)	16	0	0	0	n.a.
Peach-nectar (Spain)	0	0	16	7	2 µg/kg Paraoxon 1 µg/kg Fenhexamid 1 µg/kg Azoxystrobin
Apple Banana (Germany)	3	100	0	0	n.a.
Apple (Poland)	2	0	n.a.		n.a.
Peach in apple (Germany)	0	0	n.a.		n.a.
Apple rice (Germany)	0	0	0	0	n.a.

I<sub>ee</sub>: inhibition of electric eel AChE; I<sub>hu</sub>: inhibition of human AChE; R<sub>ee</sub>: reactivation rate of electric eel AChE; R<sub>hu</sub>: reactivation rate of human AChE; n.d.: nothing detected; n.a.: not analyzed.

Figure 1: Electric eel AChE inhibition caused by different paraoxon concentrations; direct incubation in peach infant food (black; n=7) compared with the incubation in buffer solution (gray)

Figure 2: Electric eel AChE inhibition caused by different paraoxon concentrations; direct incubation in apple infant food (black; n=6) compared with the incubation in buffer solution (gray)

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